

6,6'-Dimethoxy-2,2'-[(cyclohexane-1,2-diyl)bis(nitrilomethylidene)]diphenol

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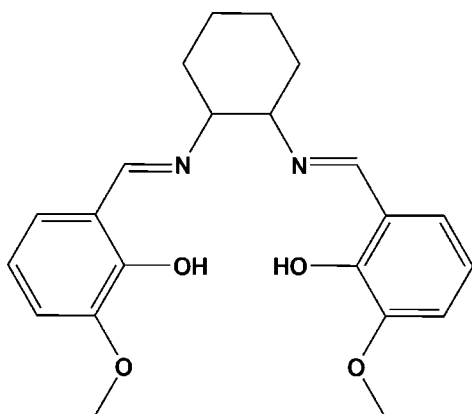
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Key indicators: single-crystal X-ray study; $T = 293$ K; mean $\sigma(\text{C}-\text{C}) = 0.004$ Å; R factor = 0.072; wR factor = 0.164; data-to-parameter ratio = 18.3.

The molecule of the title compound, $\text{C}_{22}\text{H}_{26}\text{N}_2\text{O}_4$, has two azomethine linkages, both of which are in an *E* configuration. The cyclohexyl ring adopts a chair conformation. The dihedral angle between the benzene rings is 66.57 (9)°. The molecular structure is stabilized by two intramolecular $\text{O}-\text{H}\cdots\text{N}$ hydrogen bonds.

Related literature

For related structures, see: Aslantaş *et al.* (2007); Tozzo *et al.* (2008).



Experimental

Crystal data

$\text{C}_{22}\text{H}_{26}\text{N}_2\text{O}_4$	$V = 2083.5$ (6) Å ³
$M_r = 382.45$	$Z = 4$
Monoclinic, $P2_1/c$	Mo $K\alpha$ radiation
$a = 15.014$ (3) Å	$\mu = 0.08$ mm ⁻¹
$b = 12.029$ (2) Å	$T = 293$ K
$c = 12.099$ (2) Å	$0.22 \times 0.20 \times 0.15$ mm
$\beta = 107.54$ (3)°	

Data collection

Rigaku R-Axis RAPID diffractometer	4748 independent reflections
19987 measured reflections	2352 reflections with $I > 2\sigma(I)$
	$R_{\text{int}} = 0.062$

Refinement

$R[F^2 > 2\sigma(F^2)] = 0.072$	H atoms treated by a mixture of independent and constrained refinement
$wR(F^2) = 0.164$	$\Delta\rho_{\text{max}} = 0.13$ e Å ⁻³
$S = 1.03$	$\Delta\rho_{\text{min}} = -0.18$ e Å ⁻³
4748 reflections	
259 parameters	
1 restraint	

Table 1

Hydrogen-bond geometry (Å, °).

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
$\text{O3}-\text{H3}\cdots\text{N1}$	0.88 (2)	1.77 (2)	2.572 (3)	150 (3)
$\text{O4}-\text{H4}\cdots\text{N2}$	0.86 (3)	1.80 (3)	2.587 (3)	152 (3)

Data collection: *RAPID-AUTO* (Rigaku, 1998); cell refinement: *RAPID-AUTO*; data reduction: *CrystalClear* (Rigaku/MS, 2002); program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *SHELXTL* (Sheldrick, 2008); software used to prepare material for publication: *SHELXL97*.

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Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: NG5006).

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supplementary materials

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6,6'-Dimethoxy-2,2'-[(cyclohexane-1,2-diyl)bis(nitrilomethylidyne)]diphenol

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Comment

We present the crystal structure of the title compound, as shown in Fig. 1. X-ray analysis suggests that the imino group is in the *trans* configuration and the two aromatic rings are lying on in front of the other. Bond lengths and angles within the aromatic rings are consistent with reported examples. Short H-bonds exists between the OH groups and the imino groups in *ortho* position.

Experimental

The title compound was prepared by a known method. *o*-Vanillin (2 mmol, 0.304 g) in acetonitrile (20 ml) and *trans*-1,2-cyclohexanediamine (1 mmol, 0.114 g) in methanol (20 ml) were mixed and refluxed for about 4 h at 358 K. The reaction mixture was cooled and filtered; Compound was obtained by crystallization from a mixture methanol/acetonitrile solution after a few days. Analysis: calculated for C₂₂H₂₆N₂O₄: C 69.09, H 6.85, N 7.32, O 16.73%; found: C 69.21, H 7.01, N 7.38%.

Refinement

H atoms bound to C atoms were placed in calculated positions and treated as riding on their parent atoms, with C—H = 0.93 Å (aromatic C), C—H = 0.97 Å (methylene C), and with $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C})$ or C—H = 0.96 Å (methly C) and with $U_{\text{iso}}(\text{H}) = 1.5U_{\text{eq}}(\text{C})$. The H atoms attached to the O atoms were found from the Fourier difference map and the O—H bonds are refined in the normal range with $U_{\text{iso}}(\text{H}) = 1.5U_{\text{eq}}(\text{O})$.

Figures

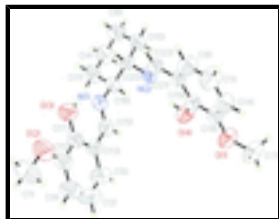


Fig. 1. The molecular structure of the title compound, showing 50% probability displacement ellipsoids.

6,6'-Dimethoxy-2,2'-[(cyclohexane-1,2-diyl)bis(nitrilomethylidyne)]diphenol

Crystal data

C₂₂H₂₆N₂O₄

$M_r = 382.45$

Monoclinic, $P2_1/c$

Hall symbol: -P 2ybc

$F(000) = 816$

$D_x = 1.219 \text{ Mg m}^{-3}$

Mo $K\alpha$ radiation, $\lambda = 0.71073 \text{ \AA}$

Cell parameters from 4748 reflections

supplementary materials

$a = 15.014 (3) \text{ \AA}$	$\theta = 3.1\text{--}27.5^\circ$
$b = 12.029 (2) \text{ \AA}$	$\mu = 0.08 \text{ mm}^{-1}$
$c = 12.099 (2) \text{ \AA}$	$T = 293 \text{ K}$
$\beta = 107.54 (3)^\circ$	Block, yellow
$V = 2083.5 (6) \text{ \AA}^3$	$0.22 \times 0.20 \times 0.15 \text{ mm}$
$Z = 4$	

Data collection

Rigaku R-Axis RAPID diffractometer	2352 reflections with $I > 2\sigma(I)$
Radiation source: fine-focus sealed tube graphite	$R_{\text{int}} = 0.062$
Detector resolution: $10.000 \text{ pixels mm}^{-1}$	$\theta_{\text{max}} = 27.5^\circ, \theta_{\text{min}} = 3.1^\circ$
ω scans	$h = -19 \rightarrow 19$
19987 measured reflections	$k = -15 \rightarrow 15$
4748 independent reflections	$l = -15 \rightarrow 14$

Refinement

Refinement on F^2	Primary atom site location: structure-invariant direct methods
Least-squares matrix: full	Secondary atom site location: difference Fourier map
$R[F^2 > 2\sigma(F^2)] = 0.072$	Hydrogen site location: inferred from neighbouring sites
$wR(F^2) = 0.164$	H atoms treated by a mixture of independent and constrained refinement
$S = 1.03$	$w = 1/[\sigma^2(F_o^2) + (0.0509P)^2 + 0.6508P]$
4748 reflections	where $P = (F_o^2 + 2F_c^2)/3$
259 parameters	$(\Delta/\sigma)_{\text{max}} = 0.001$
1 restraint	$\Delta\rho_{\text{max}} = 0.13 \text{ e \AA}^{-3}$
	$\Delta\rho_{\text{min}} = -0.18 \text{ e \AA}^{-3}$

Special details

Geometry. All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

Refinement. Refinement of F^2 against ALL reflections. The weighted R -factor wR and goodness of fit S are based on F^2 , conventional R -factors R are based on F , with F set to zero for negative F^2 . The threshold expression of $F^2 > \sigma(F^2)$ is used only for calculating R -factors(gt) etc. and is not relevant to the choice of reflections for refinement. R -factors based on F^2 are statistically about twice as large as those based on F , and R -factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

x	y	z	$U_{\text{iso}}^*/U_{\text{eq}}$
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C1	0.0604 (3)	0.6458 (3)	0.0711 (3)	0.1152 (12)
H1A	0.0594	0.6532	-0.0083	0.173*
H1B	0.1008	0.7015	0.1171	0.173*
H1C	-0.0016	0.6553	0.0765	0.173*
C2	0.0895 (3)	0.5459 (3)	0.4138 (4)	0.1002 (11)
H2A	0.0743	0.5934	0.4661	0.120*
C3	0.6339 (2)	0.4074 (3)	0.4551 (3)	0.0913 (10)
H3A	0.6771	0.4377	0.4223	0.110*
C4	0.2019 (2)	-0.1264 (2)	0.3541 (3)	0.0877 (9)
H4A	0.2139	-0.1370	0.4368	0.105*
H4B	0.1573	-0.1825	0.3140	0.105*
C5	0.1206 (2)	0.4412 (3)	0.4475 (3)	0.0897 (9)
H5A	0.1276	0.4182	0.5231	0.108*
C6	0.2916 (2)	-0.1400 (2)	0.3237 (3)	0.0832 (9)
H6A	0.2788	-0.1342	0.2404	0.100*
H6B	0.3175	-0.2132	0.3475	0.100*
C7	0.6257 (2)	0.5092 (3)	0.7826 (3)	0.1011 (11)
H7A	0.6137	0.5225	0.8550	0.152*
H7B	0.6185	0.5774	0.7395	0.152*
H7C	0.6883	0.4821	0.7971	0.152*
C8	0.5739 (2)	0.3277 (3)	0.3967 (2)	0.0801 (9)
H8A	0.5761	0.3042	0.3243	0.096*
C9	0.0802 (2)	0.5821 (2)	0.3029 (3)	0.0869 (9)
H9A	0.0599	0.6542	0.2813	0.104*
C10	0.36190 (19)	-0.0523 (2)	0.3828 (2)	0.0712 (8)
H10A	0.4176	-0.0605	0.3587	0.085*
H10B	0.3796	-0.0634	0.4660	0.085*
C11	0.16120 (19)	-0.0121 (2)	0.3201 (3)	0.0805 (9)
H11A	0.1046	-0.0040	0.3421	0.097*
H11B	0.1449	-0.0038	0.2366	0.097*
C12	0.10074 (19)	0.5126 (2)	0.2240 (3)	0.0721 (8)
C13	0.14201 (18)	0.3684 (2)	0.3698 (2)	0.0645 (7)
C14	0.6317 (2)	0.4439 (2)	0.5620 (2)	0.0732 (8)
H14A	0.6731	0.4987	0.6009	0.088*
C15	0.17636 (18)	0.2574 (2)	0.4061 (2)	0.0658 (7)
H15A	0.1869	0.2370	0.4831	0.079*
C16	0.23034 (17)	0.0781 (2)	0.3785 (2)	0.0605 (7)
H16A	0.2407	0.0746	0.4624	0.073*
C17	0.13111 (18)	0.4042 (2)	0.2567 (3)	0.0677 (7)
C18	0.56880 (19)	0.3998 (2)	0.6112 (2)	0.0605 (7)
C19	0.50689 (17)	0.3178 (2)	0.5532 (2)	0.0546 (6)
C20	0.32359 (16)	0.0644 (2)	0.3545 (2)	0.0565 (6)
H20A	0.3149	0.0794	0.2723	0.068*
C21	0.50889 (17)	0.2810 (2)	0.44483 (19)	0.0558 (6)
C22	0.44426 (17)	0.1969 (2)	0.3816 (2)	0.0574 (6)
H22A	0.4441	0.1787	0.3068	0.069*
N1	0.19248 (14)	0.18712 (18)	0.33625 (18)	0.0628 (6)
N2	0.38790 (14)	0.14722 (17)	0.42452 (16)	0.0568 (5)
O1	0.56134 (14)	0.42882 (16)	0.71764 (15)	0.0826 (6)

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O2	0.09388 (17)	0.53935 (17)	0.1121 (2)	0.1010 (7)
O3	0.15024 (16)	0.33730 (17)	0.17764 (17)	0.0920 (7)
H3	0.166 (3)	0.2728 (19)	0.211 (3)	0.138*
O4	0.44555 (14)	0.27515 (16)	0.60379 (15)	0.0754 (6)
H4	0.414 (2)	0.225 (3)	0.557 (3)	0.113*

Atomic displacement parameters (\AA^2)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
C1	0.121 (3)	0.082 (2)	0.148 (3)	0.027 (2)	0.048 (3)	0.034 (2)
C2	0.113 (3)	0.078 (2)	0.125 (3)	-0.003 (2)	0.058 (2)	-0.029 (2)
C3	0.093 (2)	0.111 (3)	0.081 (2)	-0.041 (2)	0.0441 (18)	-0.012 (2)
C4	0.082 (2)	0.0693 (19)	0.112 (2)	-0.0142 (16)	0.0286 (19)	0.0015 (18)
C5	0.105 (3)	0.081 (2)	0.095 (2)	-0.0020 (19)	0.049 (2)	-0.0140 (19)
C6	0.083 (2)	0.0656 (18)	0.098 (2)	0.0009 (16)	0.0235 (18)	-0.0063 (17)
C7	0.116 (3)	0.106 (2)	0.073 (2)	-0.028 (2)	0.0178 (19)	-0.0256 (19)
C8	0.086 (2)	0.099 (2)	0.0646 (17)	-0.0261 (18)	0.0367 (16)	-0.0097 (16)
C9	0.075 (2)	0.0583 (18)	0.130 (3)	-0.0048 (15)	0.036 (2)	-0.011 (2)
C10	0.0629 (18)	0.0709 (18)	0.0779 (18)	0.0088 (15)	0.0182 (14)	0.0011 (15)
C11	0.0510 (17)	0.082 (2)	0.105 (2)	-0.0047 (15)	0.0185 (16)	0.0027 (18)
C12	0.0568 (18)	0.0676 (19)	0.089 (2)	0.0003 (14)	0.0181 (15)	-0.0017 (17)
C13	0.0553 (16)	0.0637 (17)	0.0768 (18)	-0.0028 (13)	0.0231 (14)	-0.0080 (15)
C14	0.071 (2)	0.0743 (18)	0.0709 (18)	-0.0194 (15)	0.0154 (15)	-0.0041 (15)
C15	0.0591 (17)	0.0781 (19)	0.0619 (15)	-0.0067 (14)	0.0207 (13)	-0.0010 (15)
C16	0.0526 (16)	0.0638 (16)	0.0643 (15)	0.0046 (13)	0.0163 (12)	0.0075 (13)
C17	0.0513 (16)	0.0673 (18)	0.0813 (19)	0.0062 (14)	0.0150 (14)	-0.0084 (16)
C18	0.0653 (17)	0.0625 (16)	0.0514 (14)	0.0017 (14)	0.0142 (13)	0.0037 (13)
C19	0.0539 (15)	0.0601 (15)	0.0508 (14)	0.0007 (13)	0.0175 (12)	0.0066 (12)
C20	0.0523 (15)	0.0650 (16)	0.0519 (13)	-0.0002 (13)	0.0151 (12)	0.0026 (13)
C21	0.0546 (16)	0.0651 (16)	0.0475 (13)	-0.0021 (13)	0.0151 (12)	0.0028 (12)
C22	0.0566 (16)	0.0722 (17)	0.0446 (13)	0.0030 (13)	0.0170 (12)	0.0015 (13)
N1	0.0553 (13)	0.0680 (14)	0.0620 (13)	0.0058 (11)	0.0129 (11)	-0.0021 (12)
N2	0.0513 (13)	0.0666 (13)	0.0527 (11)	-0.0026 (10)	0.0157 (10)	0.0019 (10)
O1	0.0954 (15)	0.0916 (14)	0.0620 (11)	-0.0214 (12)	0.0258 (10)	-0.0159 (11)
O2	0.1142 (18)	0.0841 (15)	0.1000 (17)	0.0303 (13)	0.0250 (14)	0.0171 (13)
O3	0.1182 (18)	0.0836 (14)	0.0708 (13)	0.0356 (13)	0.0233 (12)	0.0012 (11)
O4	0.0820 (14)	0.0921 (15)	0.0601 (11)	-0.0245 (11)	0.0335 (10)	-0.0087 (10)

Geometric parameters (\AA , $^\circ$)

C1—O2	1.410 (3)	C10—H10A	0.9700
C1—H1A	0.9600	C10—H10B	0.9700
C1—H1B	0.9600	C11—C16	1.521 (3)
C1—H1C	0.9600	C11—H11A	0.9700
C2—C5	1.363 (4)	C11—H11B	0.9700
C2—C9	1.378 (4)	C12—O2	1.365 (3)
C2—H2A	0.9300	C12—C17	1.399 (4)
C3—C8	1.358 (4)	C13—C17	1.396 (4)
C3—C14	1.377 (4)	C13—C15	1.452 (4)

C3—H3A	0.9300	C14—C18	1.367 (3)
C4—C6	1.509 (4)	C14—H14A	0.9300
C4—C11	1.510 (4)	C15—N1	1.269 (3)
C4—H4A	0.9700	C15—H15A	0.9300
C4—H4B	0.9700	C16—N1	1.459 (3)
C5—C13	1.392 (4)	C16—C20	1.522 (3)
C5—H5A	0.9300	C16—H16A	0.9800
C6—C10	1.511 (4)	C17—O3	1.346 (3)
C6—H6A	0.9700	C18—O1	1.371 (3)
C6—H6B	0.9700	C18—C19	1.392 (3)
C7—O1	1.425 (3)	C19—O4	1.351 (3)
C7—H7A	0.9600	C19—C21	1.393 (3)
C7—H7B	0.9600	C20—N2	1.466 (3)
C7—H7C	0.9600	C20—H20A	0.9800
C8—C21	1.397 (3)	C21—C22	1.450 (3)
C8—H8A	0.9300	C22—N2	1.268 (3)
C9—C12	1.371 (4)	C22—H22A	0.9300
C9—H9A	0.9300	O3—H3	0.876 (18)
C10—C20	1.516 (3)	O4—H4	0.86 (3)
O2—C1—H1A	109.5	C4—C11—H11B	109.4
O2—C1—H1B	109.5	C16—C11—H11B	109.4
H1A—C1—H1B	109.5	H11A—C11—H11B	108.0
O2—C1—H1C	109.5	O2—C12—C9	125.5 (3)
H1A—C1—H1C	109.5	O2—C12—C17	114.9 (3)
H1B—C1—H1C	109.5	C9—C12—C17	119.6 (3)
C5—C2—C9	120.6 (3)	C5—C13—C17	119.1 (3)
C5—C2—H2A	119.7	C5—C13—C15	120.5 (3)
C9—C2—H2A	119.7	C17—C13—C15	120.4 (2)
C8—C3—C14	121.0 (3)	C18—C14—C3	120.0 (3)
C8—C3—H3A	119.5	C18—C14—H14A	120.0
C14—C3—H3A	119.5	C3—C14—H14A	120.0
C6—C4—C11	110.5 (2)	N1—C15—C13	122.2 (3)
C6—C4—H4A	109.6	N1—C15—H15A	118.9
C11—C4—H4A	109.6	C13—C15—H15A	118.9
C6—C4—H4B	109.6	N1—C16—C11	109.8 (2)
C11—C4—H4B	109.6	N1—C16—C20	108.3 (2)
H4A—C4—H4B	108.1	C11—C16—C20	111.9 (2)
C2—C5—C13	120.5 (3)	N1—C16—H16A	108.9
C2—C5—H5A	119.8	C11—C16—H16A	108.9
C13—C5—H5A	119.8	C20—C16—H16A	108.9
C4—C6—C10	110.9 (2)	O3—C17—C13	121.7 (2)
C4—C6—H6A	109.5	O3—C17—C12	118.5 (3)
C10—C6—H6A	109.5	C13—C17—C12	119.7 (3)
C4—C6—H6B	109.5	C14—C18—O1	125.0 (2)
C10—C6—H6B	109.5	C14—C18—C19	120.0 (2)
H6A—C6—H6B	108.0	O1—C18—C19	115.0 (2)
O1—C7—H7A	109.5	O4—C19—C18	118.9 (2)
O1—C7—H7B	109.5	O4—C19—C21	121.1 (2)
H7A—C7—H7B	109.5	C18—C19—C21	120.0 (2)

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O1—C7—H7C	109.5	N2—C20—C10	111.1 (2)
H7A—C7—H7C	109.5	N2—C20—C16	107.46 (19)
H7B—C7—H7C	109.5	C10—C20—C16	111.4 (2)
C3—C8—C21	120.3 (3)	N2—C20—H20A	108.9
C3—C8—H8A	119.9	C10—C20—H20A	108.9
C21—C8—H8A	119.9	C16—C20—H20A	108.9
C12—C9—C2	120.5 (3)	C19—C21—C8	118.7 (2)
C12—C9—H9A	119.8	C19—C21—C22	121.2 (2)
C2—C9—H9A	119.8	C8—C21—C22	120.1 (2)
C6—C10—C20	112.1 (2)	N2—C22—C21	122.4 (2)
C6—C10—H10A	109.2	N2—C22—H22A	118.8
C20—C10—H10A	109.2	C21—C22—H22A	118.8
C6—C10—H10B	109.2	C15—N1—C16	119.8 (2)
C20—C10—H10B	109.2	C22—N2—C20	119.2 (2)
H10A—C10—H10B	107.9	C18—O1—C7	117.2 (2)
C4—C11—C16	111.1 (2)	C12—O2—C1	118.4 (3)
C4—C11—H11A	109.4	C17—O3—H3	107 (2)
C16—C11—H11A	109.4	C19—O4—H4	106 (2)

Hydrogen-bond geometry (\AA , $^\circ$)

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
O3—H3 \cdots N1	0.88 (2)	1.77 (2)	2.572 (3)	150 (3)
O4—H4 \cdots N2	0.86 (3)	1.80 (3)	2.587 (3)	152 (3)

Fig. 1

